

## PRODUCTION OF CRYSTALLIZED ZIRCONIA SOL

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- European:

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### Abstract of JP1076919

PURPOSE: To obtain the titled stable sol having fine particle size and free from secondary coagulation tendency, with simple process, by heating an aqueous suspension containing zirconium hydroxide, keeping the hot state in an acidic atmosphere having a specific pH range, evaporating and drying the obtained aqueous suspension containing crystallized ZrO<sub>2</sub> and adding the dried product to an aqueous solvent. CONSTITUTION: An aqueous solution of a Zr salt (e.g., ZrCl<sub>4</sub>) and, as necessary, salt of other element (e.g., YCl<sub>3</sub>) is neutralized and coprecipitated with a base to obtain an aqueous suspension containing zirconium hydroxide and, as necessary, hydroxide of other element such as Y. The aqueous dispersion is adjusted to a concentration of <=1 mol/l optionally after washing, heated at >=80 deg.C, adjusted to an acidic state of pH1-5.5 e.g. by the addition of an acid and maintained in hot state of e.g. 150 deg.C for about 12hr until crystallized substance is sufficiently produced. The obtained aqueous dispersion containing crystallized ZrO<sub>2</sub> is evacuated to dryness at <=200 deg.C preferably under reduced pressure at <=100 deg.C and the obtained solid lump is added to an aqueous solvent under normal temperature and pressure condition to obtain the objective sol.

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## (54) PRODUCTION OF CRYSTALLIZED ZIRCONIA SOL

### (57)Abstract:

**PURPOSE:** To obtain the titled stable sol having fine particle size and free from secondary coagulation tendency, with simple process, by heating an aqueous suspension containing zirconium hydroxide, keeping the hot state in an acidic atmosphere having a specific pH range, evaporating and drying the obtained aqueous suspension containing crystallized ZrO<sub>2</sub> and adding the dried product to an aqueous solvent.

**CONSTITUTION:** An aqueous solution of a Zr salt (e.g., ZrCl<sub>4</sub>) and, as necessary, salt of other element (e.g., YCl<sub>3</sub>) is neutralized and coprecipitated with a base to obtain an aqueous suspension containing zirconium hydroxide and, as necessary, hydroxide of other element such as Y. The aqueous dispersion is adjusted to a concentration of  $\leq 1\text{mol./l}$  optionally after washing, heated at  $\geq 80^\circ\text{C}$ , adjusted to an acidic state of pH 1W5.5 e.g. by the addition of an acid and maintained in hot state of e.g. 150°C for about 12hr until crystallized substance is sufficiently produced. The obtained aqueous dispersion containing crystallized ZrO<sub>2</sub> is evacuated to dryness at  $\leq 200^\circ\text{C}$  preferably under reduced pressure at  $\leq 100^\circ\text{C}$  and the obtained solid lump is added to an aqueous solvent under normal temperature and pressure condition to obtain the objective sol.

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